



MACHAKOS UNIVERSITY

University Examinations for 2021/2022 Academic Year

SCHOOL OF EDUCATION

DEPARTMENT OF EARLY CHILDHOOD EDUCATION & EDUCATIONAL

DEPARTMENT OF EDUCATIONAL COMMUNICATION AND

TECHNOLOGY (COMTECH)

THIRD YEAR SECOND SEMESTER EXAMINATION FOR

DIPLOMA IN EDUCATION (ARTS)

ECT 305: TEACHING METHODS-CHEMISTRY

DATE: 7/12/2021

TIME: 2:00 – 4:00 PM

INSTRUCTIONS: *Answer Question ONE and any TWO Others.*

1.
 - a) Explain how you would **introduce two careers related to Chemistry** to a Form One Class so as to stimulate their interest in chemistry. (INCLUDE strategies and resources). (6 marks)
 - b) **Explain the importance of six components of a scheme of work** in the planning process of teaching Chemistry: You may make reference to a named topic. (12 marks)
 - c) Discuss any **three Values** of Science promoted through Chemistry. (6 marks)
 - d) Distinguish between Experimentation, Imagery and Speculation as processes of generating scientific knowledge. (6 marks)
2.
 - a) Explain **four key competences developed in the practical on test for Cations in chemistry**. (8 marks)
 - b) Prepare an 80-minute Lesson Plan for a practical lesson on **Test for Cations using NaOH (aq) and NH₄OH (aq)**. Include: Al⁺³, Mg⁺², Fe⁺³, Cu⁺², Zn⁺², Pb⁺², Ca⁺². (12 marks)
3. Discuss how Chemistry contributes **to any two** of the following: (20 marks)
 - (i) **The Medical profession.**
 - (ii) **Technology and Industrial Development.**
 - (iii) **Informed Citizenship.**

4. Choose **one** of the following methods of teaching of Chemistry, and discuss the following aspects: *Its definition, steps involved in its implementation, merits and demerits.* (20 marks)

- (i) The Project Method.
- (ii) The Meta-Cognitive approach.
- (iii) The Fieldtrip/Excursion Method.
- (iv) The Group Discussion Method.
- (v) The Demonstration Method.
- (vi) The Experimental Method.

5. Write briefly about **the skills tested in EACH of the following areas** of Chemistry

Practical:

- (i) **Qualitative analysis.** (6 marks).
- (ii) **Rate of reaction.** (4 marks).
- (iii) **Volumetric analysis.** (6 marks)
- (iv) **Enthalpy change.** (4 marks)